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11.3 Reactions in Aqueous Solution 28 > You have seen that mixing solutions of two ionic compounds can sometimes result in the formation of an insoluble salt called a precipitate. •Some combinations of solutions produce precipitates, while others do not.

11.3 Reactions in Aqueous Solution - Quia

The reaction of aqueous solutions of silver nitrate with sodium chlo-ride to form solid silver chloride and aqueous sodium nitrate is a double-replacement reaction. The reaction is shown in Figure 11.11. $\text{AgNO}_3(\text{aq}) + \text{NaCl}(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{NaNO}_3(\text{aq})$ This is the way you have been writing equations involving aqueous solutions of ionic compounds.

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AP Chem - Chapter 4: Aqueous Reactions and Solution Stoichiometry. STUDY. PLAY. aqueous solutions (p.113) solutions in which water is the dissolving medium. precipitation reactions (p.113) soluble reactants yield an insoluble product. acid-base reactions (p.113) H⁺ ions are transferred between reactants.

AP Chem - Chapter 4: Aqueous Reactions and Solution ...

Assume that an aqueous solution of hydroxide ion, OH⁻, represented by unshaded spheres, is allowed to mix with a solution of an acid, H_nA, represented by gray spheres. Three possible outcomes are represented by boxes (a)-(c), where the black spheres represent Aⁿ⁻, the anion of the acid. For clarity H₂O molecules are not shown.

Quiz+ | Quiz 4: Reactions in Aqueous Solutions

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Substituent Effects on an Inverse Electron Demand Hetero Diels–Alder Reaction in Aqueous Solution and Organic Solvents: Cycloaddition of Substituted Styrenes to Di(2-pyridyl)-1,2,4,5-tetrazine. The Journal of Organic Chemistry 1996 , 61 (6) , 2001-2005.

Diels-Alder reactions in aqueous solutions. Enforced ...

