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Solutions Manual Chemistry: Matter and Change • Chapter 12 239 CHAPTER 12 SOLUTIONS MANUAL Section 12.3 Liquids and Solids pages 415–424 Section 12.3 Assessment page 424 18. Contrast the arrangement of particles in solids and liquids. The particles are closer together in solids than in liquids because of intermolecular attractions.

States of Matter

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Matter—Properties and ChangesMatter—Properties and Changes

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2. A colloid consists of two parts. A) True; B) False; Answers: A) True: A colloid is a mixture in which one substance of microscopically dispersed insoluble particles is suspended throughout another substance. 3. Which is an example of a colloid? A) Milk, B) Jelly, C) Muddy water, D) All of the above

Ethiopia Grade 12 Chemistry Unit 1: Solutions Question Answers

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