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...

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and Solutions.

Chemistry. STUDY.

PLAY. Salvation. The
process of surrounding
solute particles with
solvent particles to
form a solution. Tyndall
Effect. Concentrated
colloids that are often

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cloudy or opaque.

Agitation.

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dnkeith. Chemistry

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and Solutions. soluble.

insoluble. miscible.

immiscible. describes a

substance that can be

dissolved in a given

solvent. describes a

substance that cannot

be dissolved in a given

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**chemistry chapter
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and ...**

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Section 14.3 Factors
Affecting Solvation A
substance that
dissolves in a solvent is
soluble. Two liquids
that are

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Section 2 - Solution
Concentration. Section
2 Practice. Section 3 -
Factors Affecting
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Practice.

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**Chapter Fourteen -
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Kessel ...

Homogeneous Mixtures

. Solution: a

homogeneous mixture
of two or more

substances . 2 Parts of
a simple solution:

Solute: the component
that is dissolved (least
abundant part)

Solvent: the dissolving
agent (most abundant
part)

Aqueous Solution:

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a substance (solute) is dissolved in water (solvent) Complex Solutions: More than one solute

Chapter 14: **Mixtures & Solutions**

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A mixture in which particles can be seen and easily separated....

A mixture containing small, undissolved particles that do not....

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the chaotic movement of colloidal particles, caused by collisions.... A mixture in which particles can be seen and easily separated....

mixtures and solutions chapter 14 Flashcards and Study ...

Chemistry II: Chapter 14 Mixtures and Solutions. a type of heterogeneous mixture whose particles settle out over time and can

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be separated from the mixture by filtration. a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension

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Solutions Manual

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Change • Chapter 14
277 Section 14.1 Types
of Mixtures pages 476
- 479 Section

Assessment 14.1 page
479 1. Explain Use the
properties of seawater
to describe the
characteristics of
mixtures. Answers will
vary but might include
that

Mixtures and Solutions

Chemistry II: Chapter
14 Mixtures and
Page 15/23

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Solutions. a type of heterogeneous mixture whose particles settle out over time and can be separated from the mixture by filtration. a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles).

**Chemistry II:
Chapter 14 Mixtures
and Solutions ...**

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Chapter 14 Sec. 1-4

Mixtures and Solutions

NOTES (white) RULE:

"LIKE DISSOLVES LIKE"

S = solubility (usually
mass/volume)

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Solutions by Sydney
Sturgeon

Chapter Fourteen:
Mixtures and Solutions

Section 2: Solution

Concentration

Concentration: a

measure, in a solution,

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of how much solute is dissolved in a specific amount of solvent or solution Describe qualitatively as concentrated or dilute Percent by mass: the ratio of the solute's mass to the solutions mass expressed as a percent.

Chapter Fourteen: Mixtures and Solutions Section 2

...

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Mixtures, Taking
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