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Chapter 15 Solutions Answers Chemistry

Chapter 15 Solutions. 1. A homogeneous mixture is a combination of two (or more) pure substances that is uniform in composition and appearance throughout. Examples of homogeneous mixtures in the real world include

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rubbing alcohol (70% isopropyl alcohol, 30% water) and gasoline (a mixture of hydrocarbons). 2.

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PLAY. solution. a homogenous mixture

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that exists in a single phase. soluble.
occurs when a substance (solute) is able
to dissolve in a solvent. miscible. occurs
when two different liquids are soluble in
each other. insoluble.

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Question 2.

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Chapter 15 Some ...

Chapter 15 263 Key Ideas Answers 8.
Hydrocarbons (compounds composed of carbon and hydrogen) in which all of the carbon-carbon bonds are single bonds are called alkanes. 10. When a(n) small section of an organic molecule is largely responsible for the molecule's chemical and physical characteristics, that section

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is called a functional group.

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Identify the solute and the solvent in....

Identify the solute and the solvent in....

Water is a polar solvent; CCl₄ is a non....

Sodium chloride is the solute, water is the solvent. Water is the solute, ethanol is the solvent. Oxygen is the solute, nitrogen is the solvent. CCl₄. Identify the solute and the solvent in....

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