

## Chapter 17 Thermochemistry Study Answers

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Chapter 17 Thermochemistry183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

### **SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)**

CHAPTER17, Thermochemistry (continued) Use the heating curve for water shown below to answer

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Questions 5,6, and 7. Heating Curve for Water Boiling point ~::>,-,----,1 ~ Melting E point ~ <--\_----'I  
Heat supplied 5. Label the melting point and boiling point temperatures on the graph. 6. What happens to the temperature during melting and vaporization?

### **THERMOCHEMISTRY**

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17 Thermochemistry Answer Key CHAPTER 17,Thermochemistry (continued) 6. In thermochemical calculations, is the direction of heat flow given from the point of view of the system, or of the surroundings? 7. What universal law states that energy can neither be created nor destroyed and can always be accounted for as work, stored potential energy, or heat?

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