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is made up of linked chapters that aim to teach the reader about a certain subject.

Chapter 21 Nuclear Chemistry Section

By the end of this section, you will be able to: Identify common particles and energies involved in nuclear reactions Write and balance nuclear equations Changes of nuclei that result in changes in their atomic

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numbers, mass numbers, or energy states are nuclear reactions.

21.2 Nuclear Equations - Chemistry

AP Chemistry CHAPTER 21- Nuclear Chemistry
21.1 Radioactivity

- When nuclei change spontaneously, emitting energy, they are said to be radioactive.
- Nuclear chemistry is the study

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of nuclear reactions
and their uses.

- Nucleons are particles in the nucleus:
- p^+ : proton
- n^0 : neutron
- Atomic number is the number of p^+ .

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L. Brown; H. Eugene

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**Chapter 21 - Nuclear
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Modern Chemistry 169

Nuclearchemistry

CHAPTER 21 REVIEW

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SECTION 1 SHORT

ANSWER Answer the
following questions in
the space provided. 1.

_____ Based on the
information about the
three elementary
particles in the text,
which has the greatest

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mass? (a) the proton
(b) the neutron (c) the
electron (d) They all
have the same mass.
2.

CHAPTER 21 REVIEW
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A nuclear fuel. A
fissionable isotope
must be present in
large enough
quantities to sustain a
controlled chain
reaction. The
radioactive isotope is
contained in tubes

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called fuel rods. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

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Chapter 21 Nuclear
Chemistry. Section
21.2 Nuclear Reactions
and Energy.

Objectives: Compare
and Contrast Nuclear
fission and Fusion,
Demonstrate Equations
that Represent the
Changes that Occur
During Radioactive

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Decay, Trace the Operation and Structure of a Nuclear Reactor. The Power of the Nucleus. Nuclear reactions involve enormous energy changes.

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Section: The Nucleus 1.
b 2. a 3. d 4. b 5. c 6. c
7. a 8. c 9. d 10. b
Section: Radioactive

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Decay 1. b 2. b 3. c 4. c

5. b 6. c 7. c 8. c 9. a

10. b Section: Nuclear

Radiation 1. d 2. d 3. c

4. b 5. d 6. a 7. c 8. c 9.

d 10. a Section:

Nuclear Fission and

Nuclear Fusion 1. c 2. d

3. a 4. c 5. c 6. b 7. d 8.

d 9. b 10. b 22 Organic

Chemistry Section:

Organic Compounds 1.

c 2. a 3. c 4. c 5. b 6. d

7. d 8. b 9. a 10. d

Section: Hydrocarbons

1. d 2. a 3. b 4. c 5. c 6.

b 7

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**Answer Key Nuclear
A and B - TEACHER
RESOURCE PAGE 21**

...

Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic idea of nuclear structure, that the nucleus of an atom is composed of protons and, with the exception

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of ${}_1^1\text{H}$, neutrons.

21.1 Nuclear Structure and Stability - Chemistry

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**Chapter 21 Nuclear
Chemistry Section 3
Answers**

Figure 21.1 Nuclear chemistry provides the basis for many useful diagnostic and therapeutic methods in

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medicine, such as these positron emission tomography (PET) scans. The PET/computed tomography scan on the left shows muscle activity. The brain scans in the center show chemical differences in dopamine signaling in the brains of addicts and nonaddicts.

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