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The element in the modern periodic table are arranged according to increasing (8) , as a result of the work of (9) . This arrangement is based on number of (10) in the nucleus of an

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atom of the element. The modern form of the periodic table results in the (11) which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) Study Guide for Content Mastery Chemistry: Matter and Change Chapter 6.

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12. Physical changes change the way something looks, but the substance stays the same. Chemical changes result in a new substance. 13. law of conservation of mass. Key Terms (page 48) Chapter Review. Part A. Vocabulary Review (page 43) 1. chemical property (1/1) 2. chemical change (4/2) 3. conservation of mass (3/2) 4. physical change (4/2) 5 ...

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Chemical Reactions

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apparent changes in matter result from changes in the groupings of atoms. Democritus further believed that matter is composed of empty space through which atoms move, that different kinds of atoms come in different sizes and shapes, and that the differing properties of atoms are due to the size, shape, and movement of the atoms.

The Structure of the Atom

T182 Chemistry: Matter and Change Name Date Class Name Date Class CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY Covalent Bonding Section 9.2 Naming Molecules In your textbook, read about how binary compounds and acids are named from their formulas.

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Study Guide - Matter—Properties and Changes Author: Catherine Last modified by: Catherine Created Date: 7/23/2017 3:41:00 PM Company: HP Other titles: Study Guide - Matter—Properties and Changes

Study Guide - Matter—Properties and Changes

Chemistry Chapter 1 Matter and Change Chemistry is the study of matter- 1. its composition 2. its structure 3. its properties 4. the changes that it undergoes Study of different branches of chemistry 1. organic- contains carbon 2. inorganic- not organic- usually no carbon 3. physical- properties & changes in matter & relation to energy 4. analytical- composition 5. biochemistry- substances and processes in living things 6. theoretical- design & predict properties of new compounds- uses math & ...

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