

## Chemistry Paper 2 SL TZ1 2012 Markscheme

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### Chemistry Paper 2 SL TZ1

Chemistry SL Paper 2 TZ1 Markscheme 2 Topics: Covalent bond, Hydrogen bond, Yield Pages: 13 (2237 words) Published: March 3, 2015

### Chemistry SL Paper 2 TZ1 Markscheme 2 - 2237 Words

[2] (b) Marit arranged the values she found in Table 12 into an energy cycle.  $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$   $3 \text{ O}_2 + 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}$   $-1411 \text{ kJ mol}^{-1}$   $-286 \text{ kJ mol}^{-1}$   $-1560 \text{ kJ mol}^{-1}$  (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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8 20 2 2 2 o correct reactants and products; [2] correct coefficients; M2 can only be scored if M1 correct. (b) Local pollutant: carbon monoxide/CO / volatile organics/VOCs / nitrogen oxide/NO / (unburnt) hydrocarbons; Do not accept methane/CH<sub>4</sub>, ethane/C<sub>2</sub>H<sub>6</sub>, propane/C<sub>3</sub>H<sub>8</sub> or butane/C<sub>4</sub>H<sub>10</sub>. Global pollutant: nitrogen oxide/NO / carbon ...

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$\text{CH}_3\text{CHOH}(\text{l}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$   $-1367 \text{ kJ mol}^{-1}$   $\text{CH}_3(\text{l}) + 10 \text{ O}_2(\text{g}) \rightarrow 8 \text{ CO}_2(\text{g}) + 9 \text{ H}_2\text{O}(\text{l})$   $-5470 \text{ kJ mol}^{-1}$  (i) Calculate the mass, in g, of ethanol and octane in 1.00 g of the fuel mixture. [1].....

### CHEMISTRY Candidate session number STandaRd lEveI PaPER 2

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(CH<sub>3</sub>)<sub>2</sub>CHCH<sub>2</sub>OH; (CH<sub>3</sub>)<sub>3</sub>COH; [2 max] Accept condensed or full structural formulas. Penalise missing H atoms or incorrect bonds (such as C-HO, C-H 2C) only once in the whole paper. (c) C H CHOCH [O] C H COCH H O<sub>25</sub> 3 25 3 2+→ +; [1] Accept condensed or full structural formulas. Accept [O] on top of the arrow. Do not accept equation without H 2O.

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