Definition Of Saturated Solution

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Definition Of Saturated Solution

Updated February 03, 2020 A saturated solution is a chemical solution containing the maximum concentration of a solute dissolve in a saturated solution. The amount of solute that can be dissolved in a solvent to form a saturated solution depends on a variety of factors.

Saturated Solution Definition and Examples

saturated solution definition: 1. a solution (= a liquid containing a solid) in which as much solid as possible is dissolved 2. a.... Learn more. Cambridge Dictionary +Plus

SATURATED SOLUTION | meaning in the Cambridge English ...

saturated solution. A solution where there is an equilibrium between the solution and its solute. Dictionary of Unfamiliar Words by Diagram Group Copyright © 2008 by Diagram Visual Information Limited.

Saturated solution - definition of saturated solution by ...

A saturated solution is a solution in which there is so much solute that if there was any more, it would not dissolve. When a saturated solution.

Saturated solution definition and meaning | Collins ...

Your browser doesn't support HTML5 audio / sæts. a r.ei.tid sa'lu:. a n / a solution (= a liquid containing a solid) in which as much solid as possible is dissolved Thesaurus: synonyms and related words

SATURATED SOLUTION | definition in the Cambridge English ...

saturated solution one in which the solvent has taken up all of the dissolved substance that it can hold in solution. sclerosing solution one containing an irritant substance (sclerosing agent) that will cause obliteration of a space, as in sclerotherapy.

Saturated solution | definition of saturated solution by ...

Definition of Saturated Solution A saturated solution is one where there are about equal amounts of particles or solutes and solvent in the solution. If you live on one of the coasts, you've...

Saturated Solution: Definition & Examples - Video & Lesson ...

A saturated solution is a solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that a solution that a solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that a solution that a solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that a solution that a solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that a solution that a solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution that contains the condition at which the solution that can be dissolved under the condition at which the solution that contains the condition at which the solution that contains the condition at which the solution at which the solution that contains the condition at which the solution at which the solution that contains the condition at which the solution at which the solut

What is a Saturated Solution - Preparation, Types & Examples

saturated solution (Noun) a solution in which the solvent can dissolve no more of a specific solute at a particular temperature How to pronounce saturated solution?

What does saturated solution mean? - definitions

The term saturated solution is used in chemistry to define a solution in which no more solute can be dissolved in the solution has been achieved when any additional substance that is added results in a solid precipitate or is let off as a gas.

Examples of Saturated Solution

A saturated solution is a chemical solution that contains the highest bound solvent level. In a saturated solution, the extra solution will not dissolve. Depending on a multitude of variables, the quantity of fluid that can we can add in a solvent to create a saturated solution.

Saturated Solution - Definitions and Examples ...

Most people chose this as the best definition of saturated-solution: A solution in which the s... See the dictionary meaning, pronunciation, and sentence examples.

SATURATED-SOLUTION | 1 Definitions of Saturated-solution ...

A saturated solution is a solution that contains the maximum amount of solute that is capable of being dissolved. At 20°C, the maximum amount of NaCl that will dissolve in 100. g of water is 36.0 g. If any more NaCl is added past that point, it will not dissolve because the solution is saturated.

Saturated and Unsaturated Solutions | Chemistry for Non-Majors

See also supersaturated. 2. (of a colour) having a large degree of saturation. 3. (of a chemical compound) a. containing no multiple bonds and thus being incapable of undergoing additional reactions. a saturated hydrocarbon. b. containing no unpaired valence electrons.

Saturated definition and meaning | Collins English Dictionary

Saturated definition is - full of moisture : made thoroughly wet. How to use saturated in a sentence.

Saturated | Definition of Saturated by Merriam-Webster

Given scenarios, graphs, diagrams, or illustrations, the student will determine the type of solution such as saturated, supersaturated, or unsaturated.

Types of Solutions: Saturated, Supersaturated, or ...

Saturated Solution Definition Chemistry A saturated solution is a chemical solution containing the maximum concentration of a solute dissolved in the solvent. The additional solute will not dissolve in a saturated solution.

Saturated Solution Chemistry Definition

When a solution of a solid solute dissolved in a liquid solvent is saturated, it is in thermodynamic equilibrium. In order for crystallization to occur, the state of the system must be shifted to a nonequilibrium state in which the concentration of the solute in the solution exceeds its equilibrium concentration at the given solution conditions.

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