

## Form 2 Chemistry Questions And Answers

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QUESTION - CHEMISTRY FORM 2 CHEMICAL FAMILIES 1. Study the information in the table below and answer the questions that follow: Elemen t Atomic radius (nm) Ionic radius (nm) W 0.114 0.195 X 0.072 0.136 Y 0.133 0.216 Z 0.099 0.181 (a)Would these form part of a metallic or a non-metallic group? Explain

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FORM 2 CHEMISTRY TIME: 2 hrs INSTRUCTIONS Attempt all the question in the question in the spaces provided All the working must be clearly shown where appropriate. Negligence will be penalized. QUESTIONS 1.The table below shows some information concerning partials of certain elements. Element Atomic number Number of electrons Mass Number P 19 18 39

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FORM 2 CHEMISTRY 2013 INSTRUCTIONS TO CANDIDATES 1. Answer all the questions in the spaces provided. 2. Mathematical tables and electronic calculators may be used. 3. All working must be clearly shown where necessary 1. The table below shows the pH values of solutions J to N.

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Study.com has answers to your toughest chemistry homework questions with detailed step by step explanations. ... CaH<sub>2</sub> reacts with water to form hydrogen gas. CaH<sub>2</sub> (a) + 2H<sub>2</sub>O(l) ltrightarrow Ca ...

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Online Chemistry Questions and Answers - Basic Chemistry Part 2 (26-50) 26) The largest number of molecules is in (a) 36 g of water (b) 28 g of carbon monoxide (c) 46 g of ethyl alcohol (d) 54 g of nitrogen pentoxide Answer: A. 27) What are isobars? Answer: Same mass number but a different atomic number.

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Boy (8-10) in Chemistry Lab. Charles Thatcher. Getty Images If there had been a million dollars on the line for knowing all the answers, you would have just missed the prize. It's okay to miss a question or two. That just shows you have other interests! You have a good command of chemistry.

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**MATHEMATICS FORM 1 EXAMINATION PAPERS TERM 1 QUESTIONS AND ...**  
Download Ebook Form 2 Chemistry Questions And Answers Atomic number Number of electrons Mass Number P 19 18 39 Form 2 Chemistry Question Papers - 6614 Chemistry. Form 2. The Structure of the Atom and the Periodic Table. 1.1 The structure of the atom. When scientists started exploring matter, they realised that matter can be divided into

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