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homogeneous mixture of two or more substances uniformly dispersed throughout a single phase. suspension. a mixture in which particles of a material are more or less evenly dispersed throughout a liquid or gas. solute.

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and Kelvin scales. (Section 8.6) Convert the amount of one substance in a given reaction to the amount of another substance in the same reaction, whether the amounts are described by mass of pure substance or volume of a solution containing a given molarity of one of the substances. (Sections 10.1 and 10.3) Given an actual yield and a theoretical

Chapter 13 Gases - An Introduction to Chemistry

SECTION 3 Date CHAPTER 11 REVIEW Gases Class SHORT

ANSWER Answer the following questions in the space provided. c

c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L. For the expression $V =$ (a) increasing P

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1 0 g b am all e s = 34. Estimate 1 4 lb per burger and 800 lb per head of cattle. 5×10^{10} burgers \times 1 0 b.2 u 5 rg lb er = 1 \times

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b.2 u 5 rg lb er × 1 80 h 0 ea lb d = 35. population = 8 million
people Estimate 5 people per family. 5 8 pe m o i p ll l i e o p n e
p r e f o am ...

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