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CHAPTER 1 REVIEW Matter and Change SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a Technological development of a chemical product often (a) lags behind basic research on the same substance. (b) does not involve chance discoveries. (c) is driven by curiosity. (d) is done for the sake of learning something new.

## **1 Matter and Change - HUBBARD'S CHEMISTRY**

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$   
4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of  $C_3H_4$ ? 2 mol O 2:1 mol H 2O c. What is the mole ratio ...

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## **mc06se cFMsr i-vi**

Modern Chemistry Chapter 4 Section 1. Arrangements of Electrons in Atoms. STUDY. PLAY. electromagnetic radiation. A form of energy that exhibits wavelike behavior as it travels through space ( $3.00 \times 10^8$  m/s) electromagnetic spectrum. Combination of all the forms of electromagnetic radiation.

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1 mol. (b) divided by the mass of 1 mol.  $nRT$  (c) multiplied by 22.4 L. (d) divided by 22.4 L. For the expression  $V =$  (a) increasing P (b) decreasing T

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CHAPTER 14 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 14-1 SHORT ANSWER Answer the following questions in the space provided. 1. Use the guidelines in Table 14-1 on page 427 of the text to predict the solubility of the following compounds in water: a. magnesium nitrate b. barium sulfate c. calcium carbonate d. ammonium ...

### 14 Ions in Aqueous Solutions and Colligative Properties

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Name Date Class CHAPTER 2 REVIEW Measurements and Calculations SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. ... MEASUREMENTS AND CALCULATIONS MODERN CHEMISTRY ...

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