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MODERN CHEMISTRY CHAPTER 2 MIXED

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REVIEW 15 HRW material copyrighted under notice appearing earlier in this work. Name Date Class . MIXED REVIEW continued 6. Three students were asked to determine the volume of a liquid by a method of their choosing. Each did three trials. The table below shows the results.

2 Measurements and Calculations

CHAPTER 2 REVIEW Measurements and Calculations SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.
observation and data a. A research team records the rainfall in inches

2 Measurements and Calculations

CHAPTER 2 REVIEW Measurements and Calculations SECTION 2-3 SHORT ANSWER Answer the following questions in the space provided. 1. Report the number of significant figures in each of the following values: a. 0.002 37 g d. 64

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mL b. 0.002 037 g e. 1.3 10² cm c. 350.
J f. 1.30 10² cm 2. Write the value of the
following operations using scientific
notation:

2 Measurements and Calculations

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$2\text{N}_2(\text{g}) + 5\text{O}_2(\text{g}) \rightarrow 2\text{N}_2\text{O}_5(\text{g})$ a. Balance the equation.
 $2\text{N}_2(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{N}_2\text{O}_3(\text{g})$ 4 mol NO_2 :3 mol O_2 b. What is the mole ratio of NO_2 to O_2 ? 15.0 mol c. If 20.0 mol of NO_2 form, how many moles of O_2 must have been consumed? True d. Twice as many moles of NO_2 form as moles of N_2O are consumed. True or False? False

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e. Twice as many grams of NO₂ ...

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CHAPTER 6 REVIEW Chemical Bonding
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium

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chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

6 Chemical Bonding

CHAPTER 5 REVIEW The Periodic Law
SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

CHAPTER 1 REVIEW Matter and Change
MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or

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heterogeneous substance.
homogeneous a. sugar homogeneous d.
plastic wrap homogeneous b. iron filings
heterogeneous e. cement sidewalk
heterogeneous c. granola bar 2. For
each type of investigation, select the
most appropriate ...

1 Matter and Change - HUBBARD'S CHEMISTRY

54 CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS MODERN CHEMISTRY ...
CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2
SHORT ANSWER Answer the following questions in the space provided. 1.
Assign the oxidation number to the specified element in each of the following examples: 4 a.

7 Chemical Formulas and Chemical Compounds

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CHAPTER 11 REVIEW Gases SECTION 2
SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:
increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same

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