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central carbon in

$\text{CH}_3\text{C}\equiv\text{N}$ and the bond

angle CCN are a. sp^2 ,

180° . b. sp , 180° . c.

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sp², 120°. d. sp³, 109°.

2. Which of the following statements about an sp hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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