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One of the main reactions that occurs is the metabolism of glucose, and another reaction is the detoxification of the liver. The metabolism of glucose is one of the main factors that affect the levels of insulin that are generated by the liver. The metabolism of glucose is also related to a person's risk of developing type 2 diabetes.

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reactions. A chemical reaction is a process that changes one set of chemicals into another set of chemicals. The elements or compounds that enter into the reaction are the reactants. The elements or compounds produced by the reaction are the products. Chemical reactions involve changes in the chemical bonds that join atoms in compounds. Energy in Reactions Some chemical reactions release energy; others absorb energy.

2.4 Chemical Reactions and Enzymes

Copyright © 2011 Pearson Education, Inc. - 2 - 7. Consider the following reaction: $C_6H_{12}O_6 + C_6H_{12}O_6 \rightarrow C_{12}H_{22}O_{11} + H_2O$ a. The equation is not balanced; it is missing a molecule of water. Write it in on the correct side of the equation. $C_6H_{12}O_6 + C_6H_{12}O_6 \rightarrow C_{12}H_{22}O_{11} + H_2O$ b.

Chapter 5: The Structure and Function of Large Biological ...

A chemical reaction can be concisely represented by a chemical equation. 1. The substances that undergo a chemical change are the reactants. 2. The new substances formed in a chemical reaction are the products. 3. In accordance with the law of conservation of mass, a chemical equation must be balanced. When balancing an equation, you place coefficients in front of reactants and products so that the number of atoms of each element is the same on both sides.

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Pearson Education Chemistry Answers Chemical Reactions 3 have to balance the charges of the two ions so they are equal. Introduction to Limiting Reactant and Excess Reactant Limiting reactant is also called limiting reagent. The limiting reactant or limiting reagent is the first reactant to get used up in a reaction.

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Course of Reaction Chemicals Definition Reactants The elements or compounds that enter into a chemical reaction Products The elements or compounds produced by a chemical reaction CHEMICALS IN A CHEMICAL REACTION 3. Chemical reactions always involve changes in chemical energy. Energy in Reactions (page 50) 4. What is released or absorbed whenever ...

Section 2-4 Chemical Reactions and Enzymes

$A + B \rightarrow C + D$ 1 The dependence of the reaction rate on the concentration of each reactant is given by the equation called the rate law $\text{rate} = k[A]^m[B]^n$ where k is a proportionality constant called the rate constant. The exponent m determines the reaction order with respect to A , and n determines the reaction order with respect to B .

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One sign that a chemical reaction has occurred is the formation of new substances. ____ 17. A chemical equation shows a chemical reaction using symbols instead of words. ____ 18. The reactants are the new materials produced during a chemical reaction. ____ 19. Increasing the surface area of the reactants will speed up a chemical reaction ...

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