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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $Q_1 Q_2 / 4\pi\epsilon_0 r$ First we consider the interaction between Na^+ ion and $6Cl^-$ ions at distance r Applying Coulomb's law we have $Q_1 Q_2 (-6e)(+e) / 4\pi\epsilon_0 r = -6e^2 / 4\pi\epsilon_0 r$ Similarly, we now consider 12 Na^+ ions as second nearest neighbors at a distance r $Q_1 Q_2 (+12e)(+e) / 4\pi\epsilon_0 r = 12e^2 / 4\pi\epsilon_0 r$ and Cl^- ...

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